

(12) **UK Patent Application** (19) **GB** (11)

2 167 446 A

(43) Application published 29 May 1986

(21) Application No **8524689**

(22) Date of filing **7 Oct 1985**

(30) Priority data

(31) **8425282**

(32) **5 Oct 1984**

(33) **GB**

(51) INT CL⁴
C25D 15/00

(52) Domestic classification
C7B 120 314 321 411 701 719 724 727 751 BPH
U1S 2036 C7B

(56) Documents cited
GB A 2083076

(58) Field of search
C7B
Selected US specifications from IPC sub-class C25D

(71) Applicants
BAJ Limited
(United Kingdom)
Banwell Weston-super-Mare, Avon BS24 8PD
The Secretary of State for Defence
(United Kingdom)
Whitehall, London SW1A 2HB

(72) Inventors
John Foster
Francis John Honey
Eric Charles Kedward
James Edward Restall

(74) Agent and/or Address for Service
Kilburn & Strobe, 30 John Street, London WC1N 2DD

(54) Electrode deposited composite coating

(57) A substrate such as a single crystal gas turbine blade is given a protective coating by composite electrodeposition of a metal matrix M₁ containing particles of CrAlM₂, where M₁ is Ni or Co or Fe and M₂ is Y, Si, Ti, or a rare earth element. The preferred matrix is Ni or Co or NiCo and the preferred particles are CrAlY.

The particle size is important and preferably 75% by weight are between 3.9 and 10.5µm. the coating is preferably heat treated to produce interdiffusion between the constituents of the matrix and the particles.

GB 2 167 446 A

2167446

1/2

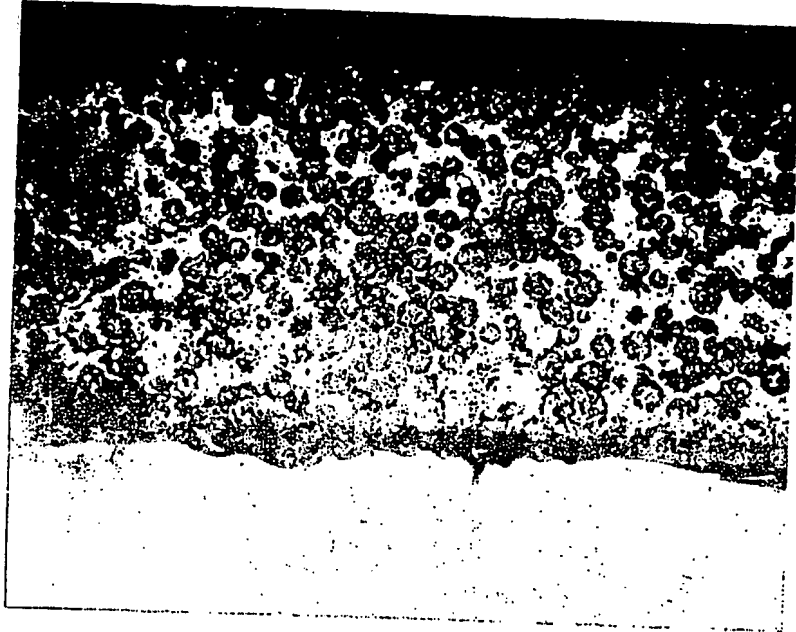


FIG. 1.

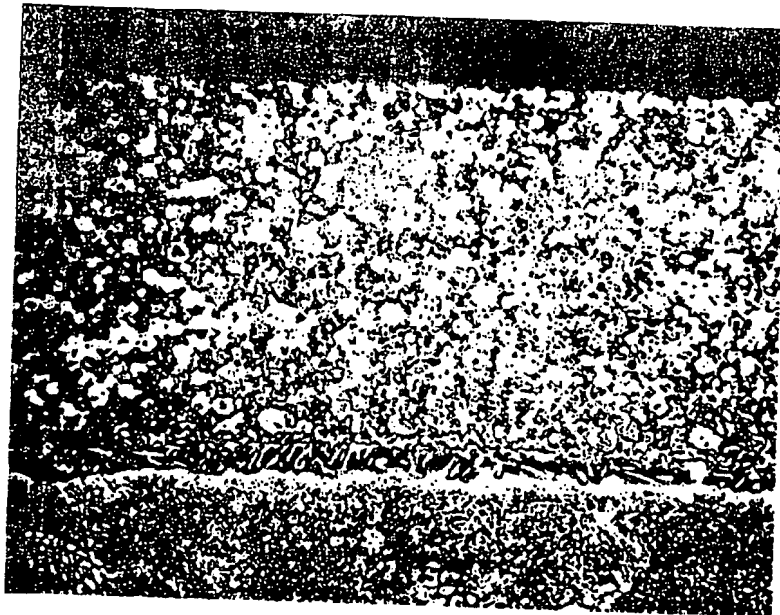


FIG. 2.

2167446

2/2

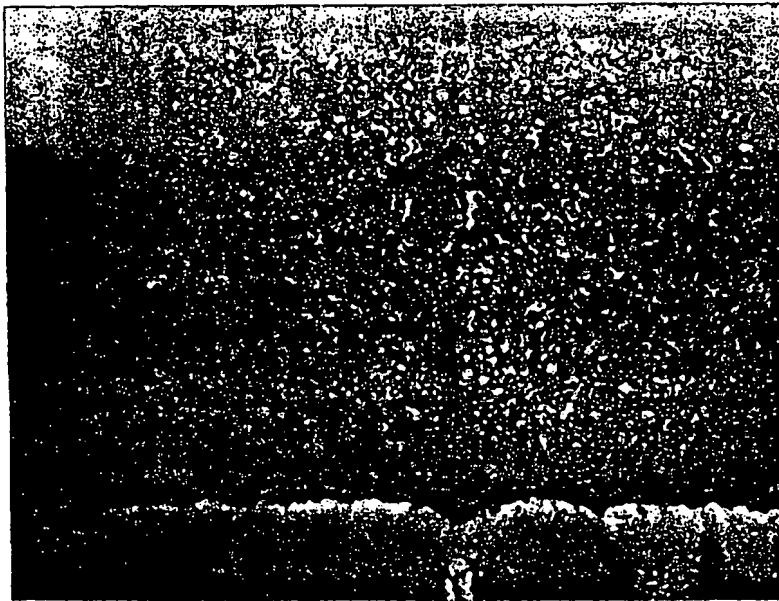


FIG. 3.

SPECIFICATION

Metallic protective coatings

- 5 This invention relates to metallic protective coatings suitable for the protection of substrates intended to operate in hostile environments to provide improved resistance to one or more of oxidation, corrosion and erosion. The invention also relates to processes for the
- 10 production of such coatings. Such coatings may be used for various purposes but are particularly valuable for the protection of high temperature aerofoil components such as rotor blades and stator vanes of gas turbines. Such coatings are employed to enable
- 15 materials for the components to be selected with greater consideration for their structural characteristics and lesser consideration for their resistance to corrosion and the like. Examples of such materials are the so-called super-alloys and particularly un-
- 20 idirectional solidified and single crystal components.

- There is described in EP-B1-0 024 802 a gas turbine blade fabricated from a nickel-based alloy which is provided by the plasma arc spraying technique with an overlay coating of CoCrAlY. This coating may suffer
- 25 from microporosity and is therefore given a further coating of aluminium by a pressure pulse chemical vapour deposition process. This two stage process has been found to give good results but is expensive and it is thought that it is not in commercial use. Two
- 30 processes for the production of overlay coatings which are used for the purpose described are plasma spraying and electron beam techniques. However, these are both expensive.

- It is accordingly an object of the present invention to
- 35 provide a coating which has good properties while being reasonable in price.

- This invention is believed to reside in one of the features set out in the immediately following paragraphs or in one of the novel features included in the
- 40 later parts of the this specification relating to more detailed examples or to any compatible combinations from either one or from both of the two aforementioned groups of features and it is to be emphasised that such combinations have been envisaged and are
- 45 intended to be disclosed hereby, even though not all of them are at present claimed.

- According to one aspect of the present invention a coating is provided by composite electro-deposition, on a substrate, of a metal matrix M_1 containing
- 50 particles of $CrAlM_2$ where M_1 is Ni or Co or Fe or two or all of these metals and M_2 is Y, Si, Ti, a rare earth element or two or more of these elements. Preferably M_1 is Ni or Co or NiCo. The electrodeposition may be electrolytic or electroless. Examples of apparatus and
- 55 processes which may be used are described in UK-A-1 218 179, 1 329 081 and 1 347 184, it being preferred to use those described in UK-B-2 014 189. The preferred particles are CrAlY. The particles could also contain one or more of Hf, Ta, Nb, Mn and Pt.

- 60 By electrodeposition there can be produced a coating in which particles forming one phase are dispersed in a matrix forming a second phase and

- such a coating has very desirable properties, and surface finish. The composite coatings may be contrasted with those produced by spraying techniques.

- The coatings will in practice be heat treated to improve their properties. An effect of heat treatment is to produce diffusion between the matrix and the particles so that the matrix of M_1 also contains some
- 70 Cr, Al and M_2 while the particles also contain some M_1 . For some applications, it is possible to treat the coating to a hot isostatic pressing (HIP) cycle to remove any traces of porosity in the coating. For example, where the matrix M_1 is Ni, the coating and, of
- 75 course, its substrate may be given a HIP cycle of 2 hours at 1120°C in argon at 140 MPa.

- It has been found that outstandingly good results are obtained by the close control of the particle size distribution. Thus, it has been found that for the best
- 80 results particles in the as-deposited coating large than about 15µm should be avoided so far as is practicable and it is therefore preferred that at least 99% of the particles co-deposited in the coating are below 25µm. An additional or alternative preference is that at least
- 85 97% of the co-deposited particles are below 15µm. A further additional or alternative preference is that at least 84% of the co-deposited particles are below 10.5µm. Particularly preferred distributions include: more than 95% by weight with particle sizes between
- 90 3.0 and 13.6µm, more than 75% between 3.9 and 10.5µm, and more than 65% between 5.0 and 10.5µm, these limits being applicable individually or cumulatively and separately or in combination with one or more of the immediately preceding preferences for
- 95 the proportions below 25µm, 15µm and 10.5µm respectively. For most conditions of co-deposition, these as-deposited limits will be achieved by applying the same limits to the size distribution of the particles in the bath but for this to be the case care must be
- 100 taken that adequate means are employed to ensure even distribution of the particles throughout the bath.

- It has been mentioned above that the coating is preferably heat treated to obtain interdiffusion between the constituents of the matrix and the particles.
- 105 While diffusion of more than one of the constituents contributes to the excellent properties of the coatings of the invention, it is through that diffusion of Al from the particles into the matrix is particularly valuable since the availability of Al at the surface of the coating
- 110 will lead to the formation of aluminium oxides at the surface when the coated component is in use and aluminium oxides are known to provide excellent erosion and corrosion resistance in hostile environments such as those encountered by the blades and
- 115 vanes of gas turbines. Preferably, the temperatures and durations of the heat treatment are selected to produce a minimum of 3 weight percent (w/o) of Al in the matrix phase of the coating and high Al values may be beneficial up to values approaching the maximum
- 120 amount of Al which is soluble in the matrix phase (thought to be about 4.2 w/o for Al in Co).

- To achieve adequate diffusion, it is probable that the lowest effective temperature will be 700°C but to achieve rapid diffusion high temperatures are prefer-

The drawing(s) originally filed was (were) informal and the print here reproduced is taken from a later filed formal copy.

red, one suitable range being between 1050°C and 1150°C, that between 1100° and 1140°C being particularly preferred, the necessary duration being between one half and two and one half hours. Very good results have been achieved with one to two hours at 1100°C in vacuum, but longer times may be beneficial in some cases and in particular if lower temperatures are used.

The proportions of the constituents of the particles may vary within wide limits but preferably the proportions of Cr and Al range between 80/20 and 50/50 while the content of M_2 in the particles will be substantially less than the other two constituents, preferably less than 2.5 w/o, preferably less than 2.0 w/o. Particularly preferred proportions are Cr-37.3 w/o Al 0 1.7 w/o Y. Although CoCrAlY coatings are suitable for many application, for example for aero engines, those of NiCoCrAlY and CoNiCrAlY may be preferred for these and other applications. For marine and industrial power plants coatings containing high percentages of chromium (e.g. up to 40 weight per cent) and with silicon (e.g. up to 10 weight per cent), platinum (e.g. up to 10 weight per cent) and hafnium (e.g. up to 5 weight per cent) may be required.

It will now be appreciated that the preferred features of the invention may be characterised in various ways. Thus according to one aspect (and realising that the invention has many broader aspects), the invention may be seen to reside in a substrate carrying an overlay coating of M_1CrAlM_2 (where M_1 is Co, Ni or NiCo and M_2 is Y, Si, Ti or two or all of these but preferably Y) applied by electrodeposition of a matrix of M_1 and particles of $CrAlM_2$ having in the as-deposited state more than 97% by weight of the particles below 15 μ m, the coating having been heated treated to produce a minimum of 3 weight percent of Al in the matrix.

The invention may be carried into practice in various ways but three Examples will now be given.

Rods of MarMOQ2 were coated by the techniques and apparatus described in UK-A-2 014 189.

In Example A, the bath contained a CoNi plating solution and the particles were of CrAlY containing 60 parts by weight Cr to 40 parts Al and 1.7 w/o Y with a size distribution, both in the bath and in the as-deposited coating, as given in column A of the following Table.

TABLE				
COATING		A	B	C
SIZE BAND		Weight	Per Cent	
118.4	54.9	0	0	0
54.9	33.7	0.2	0	0
33.7	23.7	2.9	0.3	0.1
23.7	17.7	5.9	1.3	0.7
17.7	13.6	14.8	4.3	2.9
13.6	10.3	20.2	17.7	11.7
10.3	8.2	28.1	38.1	30.5
8.2	6.4	12.4	18.3	21.1
6.4	5.0	7.8	12.3	18.0
5.0	3.9	7.5	0.2	9.8
3.9	3.0	0.2	0.1	5.1
3.0	2.4	0	0	0.1
2.4	1.9			0

After plating, the rod with the composite overlay coating thereon was given a heat treatment of one hour at 1100°C in vacuum. A micrograph (at an

original magnification of X400) of the coating after heat treatment is shown in Figure 1. The coating comprises a matrix of CoNi into which amounts of Cr, Al and Y have diffused and particles into which amounts of Co and Ni have diffused. The matrix has approximately 20% Cr and approximately 4% Al. The coating has good properties.

In Example B, the bath contained a Co plating solution and the particles were of CrAlY containing 50 parts by weight Cr to 50 parts by weight of Al and 1.7 w/o Y with a size distribution, both in the bath and in the as-deposited coating, as given in Column B in the above Table. The rod was heat treated as for Example A and a micrograph of the coating after heat treatment is shown in Figure 2. The coating has good properties. In a test, the coating was found to last in excess of 600 hours compared with a currently used aluminide coating which lasted only 200 hours.

In Example C, the bath contained a Co plating solution and the particles were of CrAlY containing 60 parts by weight Cr to 40 Parts Al and 1.7 w/o Y with a size distribution, both in the bath and in the as-deposited coating, as given in Column C of the above Table. The rod was heat treated as for Examples A and B and a micrograph of the coating is shown in Figure 3. It has been found that while the properties of the coatings of Examples A and B are good, those of Example C are outstanding. The coating is very dense, adherent and smooth with, as can clearly be seen from Figure 3, a fine microstructure with an even distribution of particles and no zones of weakness.

Substantial quantities of Al (as well as some Cr and Y) have diffused out of the particles into the Co matrix and Co as diffused into the particles so the coating consists of a 2-phase alloy, the matrix phase containing cobalt chromium aluminium and yttrium and the particles contain the same constituents but in different proportions. Upon oxidation at 1000°C this alloy developed an oxide on the surface which was rich in aluminium oxide and which was particularly resistant to erosion and corrosion. In a test on a burner rig test at 1100°C, the coating was found to last over 600 hours compared to coating of similar overall composition produced by argon shrouded plasma spraying which only survived 400 hours.

It will now be appreciated that the invention provides overlay coatings having extremely beneficial properties. It should be noted that the coatings employ fine particles which are evenly distributed in an equiaxed matrix, thus producing a very high quality surface finish requiring little or no additional work; in general, the coating will be given a heat treatment but no other finishing treatment. This may be contrasted with sprayed and physical vapour deposition processes which put down coatings tending to have structures which produce coarse finishes requiring very lengthy finishing. These coatings usually require peening to eliminate comprehensive stresses in the surface and to encourage recrystallisation on heat treatment. Plasma sprayed deposits normally also require a polishing operation because of the rough surface produced by this process.

CLAIMS

1. A coating produced by composite elec-

trodeposition, on a substrate, of a metal matrix M_1 containing particles of $CrAlM_2$, where M_1 is Ni or Co or Fe or two or all of these metals and M_2 is Y, Si, Ti, a rare earth element or two or more of these elements.

- 5 2. A coating as claimed in Claim 1 in which M_2 is Y.
3. A coating as claimed in Claim 1 or Claim 2 in which the particles also contain one or more of Hf, Ta, Nb, Mn and Pt.
4. A method of producing a coating on a substrate
- 10 which comprises electrodepositing a metal matrix M_1 from a bath containing particles of $CrAlM_2$ so as to codeposit the particles with the matrix, M_1 being Ni or Co or Fe or two or all of these metals and M_2 being Y, Si, Ti, a rare earth element or two or more of these
- 15 elements.
5. A method as claimed in Claim 4 in which M_2 is Y.
6. A method as claimed in Claim 4 or Claim 5 in which the particles also contain one or more of Hf, Ta,
- 20 Nb, Mn and Pt.
7. A method as claimed in any of Claims 4 to 6 in which at least 99% by weight of the particles in the as-deposited coating are below $25\mu m$.
8. A method as claimed in any of Claims 4 to 7 in
- 25 which at least 97% by weight of the particles in the as-deposited coating are below $15\mu m$.
9. A method as claimed in any of Claims 4 to 8 in which at least 84% by weight of the particles in the as-deposited coating are below $10.5\mu m$.
- 30 10. A method as claimed in any of Claims 4 to 9 in which at least 95% by weight of the particles are between 3.0 and $13.6\mu m$.
11. A method as claimed in any of Claims 4 to 10 in which at least 75% by weight of the particles are
- 35 between 3.9 and $10.5\mu m$.
12. A method as claimed in any of Claims 4 to 11 in which at least 65% by weight of the particles are between 5.0 and $10.5\mu m$.
13. A method as claimed in any of Claims 4 to 12
- 40 which includes heat treating the as-deposited coating to obtain interdiffusion between the constituents of the matrix and the particles.
14. A method as claimed in Claim 13 in which the heat treatment produces a coating having a mini-
- 45 mum of 3 weight percent of Al in the matrix phase.
15. A method as claimed in Claim 13 or Claim 14 in which the heat treatment is effected at temperatures between $1050^\circ C$ and $1150^\circ C$.
16. A method as claimed in Claim 13 or Claim 14
- 50 in which the heat treatment is effected at temperatures between $1090^\circ C$ and $1140^\circ C$.
17. A method as claimed in Claim 13 or Claim 14 in which the heat treatment is effected at $1100^\circ C$ in vacuum for between one and two hours.
- 55 18. A method as claimed in any of Claims 4 to 17 in which the proportion of Cr and Al in the as-deposited particles is between 80/20 and 50/50.
19. A method as claimed in any of Claims 4 to 18 in which the proportion of M_2 in the as-deposited
- 60 particles is less than 2.5 weight percent.
20. A method as claimed in any of Claims 4 to 18 in which the proportion of M_2 in the as-deposited particles is less than 2.0 weight percent.
21. A method as claimed in any of Claims 4 to 18 in
- 65 which the as-deposited particles consist of 37.3

weight percent Al, 1.7 weight percent Y and the balance Cr.

22. A substrate carrying an overlay coating of M_1CrAlM_2 (where M_1 is Co, Ni or NiCo and M_2 is Y, Si,
- 70 Ti or two or all of Y, Si and Ti) applied by electro-codeposition of a matrix of M_1 and particles of $CrAlM_2$ having in the as-deposited state more than 97% by weight of the particles below $15 m$, the coating having been heat treated to produce a
- 75 minimum of 3 weight percent of Al in the matrix.

23. A method of producing a coating on a substrate substantially as described herein with reference to the foregoing Example A or Example B or Example C.

Printed in the United Kingdom for Her Majesty's Stationery Office, 8818935, 5/86 18996. Published at the Patent Office, 25 Southampton Buildings, London WC2A 1AY, from which copies may be obtained.